

Izotaurcky' dej - $p = \text{konst}$ $dp = 0$

$$dQ = dW + dA'$$

$$dQ = m c_v dT + p dV$$

$$Q = m c_v (T_2 - T_1) + p (V_2 - V_1)$$

$$pV = nRT \Rightarrow V = \frac{nRT}{p}$$

$$a) Q = m c_v (T_2 - T_1) + p \left(\frac{nRT_2}{p} - \frac{nRT_1}{p} \right) = m c_v (T_2 - T_1) + nR(T_2 - T_1) =$$

$$= m c_v (T_2 - T_1) + nR(T_2 - T_1) = n(T_2 - T_1)(c_v + R)$$

Mayerova rovnici $c_p - c_v = R \Rightarrow c_p = c_v + R$

$$Q = n(T_2 - T_1) c_p = m \kappa c_v (T_2 - T_1) \quad \kappa = \frac{c_p}{c_v} \quad (493)$$

$$b) Q = m c_v (T_2 - T_1) + p (V_2 - V_1) \quad pV = nRT \Rightarrow nT = \frac{pV}{R}$$

$$Q = \frac{c_v p}{R} (V_2 - V_1) + p (V_2 - V_1) \quad nT_2 = \frac{pV_2}{R}$$

$$Q = p (V_2 - V_1) \left[\frac{c_v}{R} + 1 \right] = \frac{p}{R} (V_2 - V_1) (c_v + R) =$$

$$= \frac{p c_p}{R} (V_2 - V_1) = \frac{\kappa c_v}{R} p (V_2 - V_1) = \frac{\kappa c_v}{c_p - c_v} p (V_2 - V_1)$$

$$= \frac{\kappa c_v}{\kappa - 1} p (V_2 - V_1) = \frac{\kappa p (V_2 - V_1)}{\kappa - 1} \quad (380)$$

Adiabaticky' dej:

$$Q = \text{konst} \quad dQ = 0 \quad dA' = -dW \quad \text{práci plynu ne uloví smut. energii}$$

$$a) A' = - \int_1^2 dW = W_1 - W_2 = m c_v (T_1 - T_2)$$

$$b) A = \int_{V_1}^{V_2} p dV \quad pV^\kappa = p_1 V_1^\kappa \Rightarrow p = \frac{p_1 V_1^\kappa}{V^\kappa}$$

$$A = p_1 V_1^\kappa \int_{V_1}^{V_2} \frac{1}{V^\kappa} dV = p_1 V_1^\kappa \left[\frac{V^{1-\kappa}}{1-\kappa} \right]_{V_1}^{V_2} = \frac{p_1 V_1^\kappa}{1-\kappa} \left(\frac{V_2^{1-\kappa}}{\kappa} - V_1^{1-\kappa} \right)$$

1. veta termodyn

$$dQ = dW + dA'$$

$$dQ = \mu C_v dT + p dV$$

$$dQ = m c_v dT + p dV$$

$$\mu = \frac{m}{M_{mol}}$$

$$C_v = M_{mol} c_v$$

Izochoricky' dej $V = \text{const}$

a) $dV = 0$ $dA' = 0$

$$dQ = dW$$

$$Q = \int_1^2 dW = W_2 - W_1 = \mu c_v (T_2 - T_1)$$

Izotermicky' dej $T = \text{const}$

$$dT = 0$$

$$dW = 0$$

$$dQ = dA'$$

$$Q = \int_1^2 dA' = \int_1^2 p dV =$$

$$p_0 V_0 = p_1 V_1$$

$$p_1 = \frac{p_0 V_0}{V_1}$$

$$= p_0 V_0 \int_1^2 \frac{1}{V} dV = p_0 V_0 \ln \frac{V_2}{V_1} = p_0 V_0 (\ln V_2 - \ln V_1)$$

$$= p_0 V_0 \ln \frac{V_2}{V_1}$$

$$p_0 V_0 = p_1 V_1$$

$$\frac{V_1}{V_0} = \frac{p_0}{p_1}$$

$$pV = \mu RT$$

$$Q = \mu RT \ln \frac{V_2}{V_1} = \mu RT \ln \frac{p_0}{p_1}$$

382, 416, 417